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APPLICATION NO.		FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/446,641			TSUYONOBU HATAZAWA	P99.2641	2680
26263	7590	09/05/2002			
<del>-</del> · · · ·	_	NATH & ROSEN	EXAMINER		
P.O. BOX 06 WACKER D	RIVE S'		DOVE, TRACY MAE		
CHICAGO, IL 60606-1080				ART UNIT	PAPER NUMBER
				1745	17
				DATE MAILED: 09/05/2002	16

Please find below and/or attached an Office communication concerning this application or proceeding.

# Office Action Summary

Application No. 09/446,641

Applicant(s)

Hatazawa et al.

Examiner

**Tracy Dove** 

Art Unit 1745



	on the cover sheet with the correspondence address
Period for Reply	
A SHORTENED STATUTORY PERIOD FOR REPLY IS SET THE MAILING DATE OF THIS COMMUNICATION.	
<ul> <li>Extensions of time may be available under the provisions of 37 CFR 1.136 (a). In mailing date of this communication.</li> </ul>	n no event, however, may a reply be timely filed after SIX (6) MONTHS from the
<ul> <li>If the period for reply specified above is less than thirty (30) days, a reply within</li> <li>If NO period for reply is specified above, the maximum statutory period will apply</li> </ul>	
<ul> <li>Failure to reply within the set or extended period for reply will, by statute, cause</li> <li>Amy reply received by the Office later than three months after the mailing date of earned patent term adjustment. See 37 CFR 1.704(b).</li> </ul>	the application to become ABANDONED (35 U.S.C. § 133).
Status	
1) X Responsive to communication(s) filed on Mar 14,	2002 .
2a) ☐ This action is <b>FINAL</b> . 2b) ☒ This ac	ction is non-final.
3) Since this application is in condition for allowance closed in accordance with the practice under Ex particle.	except for formal matters, prosecution as to the merits is arte Quayle, 1935 C.D. 11; 453 O.G. 213.
Disposition of Claims	
4) 💢 Claim(s) <u>10-19</u>	is/are pending in the application.
4a) Of the above, claim(s)	is/are withdrawn from consideration.
5)  Claim(s)	is/are allowed.
6) 💢 Claim(s) <u>10-19</u>	is/are rejected.
7) 🗆 Claim(s)	is/are objected to.
8)	are subject to restriction and/or election requirement.
Application Papers	
9) $\square$ The specification is objected to by the Examiner.	
10) The drawing(s) filed on is/ar	e a) $\square$ accepted or b) $\square$ objected to by the Examiner.
Applicant may not request that any objection to the	drawing(s) be held in abeyance. See 37 CFR 1.85(a).
11) The proposed drawing correction filed on	is: a) $\square$ approved b) $\square$ disapproved by the Examiner.
If approved, corrected drawings are required in reply	to this Office action.
12) $\square$ The oath or declaration is objected to by the Exam	niner.
Priority under 35 U.S.C. §§ 119 and 120	
13) 🗓 Acknowledgement is made of a claim for foreign	priority under 35 U.S.C. § 119(a)-(d) or (f).
a) ☑ All b) ☐ Some* c) ☐ None of:	
1. Certified copies of the priority documents ha	ve been received.
2. Certified copies of the priority documents ha	ve been received in Application No
3.  Copies of the certified copies of the priority application from the International Bur *See the attached detailed Office action for a list of the second	
a) ☐ The translation of the foreign language provision	
15)☐ Acknowledgement is made of a claim for domesti	
Attachment(s)	,
1) X Notice of References Cited (PTO-892)	4) Interview Summary (PTO-413) Paper No(s).
2) Notice of Draftsperson's Patent Drawing Review (PTO-948)	5) Notice of Informal Patent Application (PTO-152)
3) Information Disclosure Statement(s) (PTO-1449) Paper No(s).	6) Other:

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#### **DETAILED ACTION**

This Office Action is in response to the communication filed on 3/14/02. Applicant's arguments have been considered, but are moot in view of the new grounds of rejection. Claims 10-19 are rejected in view of the prior art.

#### Continued Examination Under 37 CFR 1.114

A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 3/14/02 has been entered.

#### Information Disclosure Statement

The information disclosure statement filed 4/3/00 has been considered.

Claim Rejections - 35 USC § 112

All 35 U.S.C. 112 rejections have been withdrawn.

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### Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (e) the invention was described in-
- (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effect under this subsection of a national application published under section 122(b) only if the international application designating the United States was published under Article 21(2)(a) of such treaty in the English language; or
- (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that a patent shall not be deemed filed in the United States for the purposes of this subsection based on the filing of an international application filed under the treaty defined in section 351(a).

Claims 1-19 are rejected under 35 U.S.C. 102(e) as being anticipated by Gao et al., US 5,756,230.

Gao teaches a method of improving the structural integrity of an electrode binder and a polymeric matrix component of an electrochemical cell by employing polymer blends comprising fluoropolymers. See abstract. With the inventive fluoropolymer blends of Gao, the polymer binders of the anode and cathode and the polymeric layer of the electrolyte (solid) do not become brittle and crack under stress. See col. 2, lines 1-38. The fluoropolymer blends are described in col. 4, lines 19-67. The individual polymers of the blend may be homopolymers having a molecular weight in the range of 50,000 to 900,000, copolymers having a molecular weight in the range of 10,000 to 900,000 or terpolymers having a molecular weight in the range of 10,000 to 900,000. Note polytetrafluoroethylenes and polyvinyl fluorides are preferred homopolymers and

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polyvinylidene fluoride-hexaflurorpropylene is a preferred copolymer. For blends comprising a homopolymer and a copolymer, the relative weight percentage of the homopolymer preferably ranges from about 90% to 50%. See col. 5, lines 5-23. The cathode may comprise a lithium transition metal oxide and the anode may comprise carbon (col. 5, lines 59-65). Lithium ion cells are rechargeable. Lamination causes the polymeric components of the anode and cathode precursors to adhere to the polymeric layer (Example 3). The electrochemical cell includes an electrolytic solvent such as an organic carbonate (col. 5, lines 36-67).

Thus the claims 10 and 12-19 are anticipated.

Regarding claim 11, Gao teaches a fluoropolymer blend of a homopolymer having a molecular weight in the range of 50,000 to 900,000 and a copolymer having a molecular weight in the range of 10,000 to 900,000. Gao further teaches a polymer blend of a homopolymer having a molecular weight in the range of 50,000 to 900,000 and a terpolymer having a molecular weight in the range of 10,000 to 900,000. See col. 4, lines 44-65 and col. 5, lines 6-23). Thus, Gao anticipates claim 11.

Claims 10 and 12-19 are rejected under 35 U.S.C. 102(e) as being anticipated by Liu et al., US 6,096,101.

Liu teaches a polymeric structure suitable for an anode, a cathode and/or a solid electrolyte of an electrochemical cell. See abstract. The polymer is preferably a fluoropolymer such as polyvinylidene fluoride (PVDF), polyvinylidene fluoride-co-hexafluoropropylene and

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mixtures thereof (col. 2, lines 52-54). The PVDF is about 70-99.9% of the copolymer. Preferred molecular weights of the fluoropolymers are from 10,000 to 700,000 (col. 5, lines 31-41). The cathode material may be a lithium transition metal oxide (col. 8, lines 17-26) and the anode

Thus the claims are anticipated.

material may be an intercalation carbon material (col. 9, lines 44-55).

Response to Arguments

Applicant's arguments with respect to claims 10-19 have been considered but are moot in view of the new ground(s) of rejection.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Tracy Dove whose telephone number is (703) 308-8821. The Examiner may normally be reached Monday-Thursday (9:00 AM-7:30 PM). My supervisor is Pat Ryan, who can be reached at (703) 308-2383. The Art Unit receptionist can be reached at (703) 308-0661 and the official fax numbers are 703-872-9310 (after non-final) and 703-872-9311 (after final).

August 29, 2002

CAROL CHANEY
PRIMARY EXAMINER

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